# Essential Oils—Area Percent Calculations Using Deconvoluted Total Ion Chromatogram 

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## Key Words: GC-TOFMS

## 1. Introduction

When using a mass spectrometer as a detector for gas chromatography to obtain what is called a Total Ion Chromatogram (TIC), it is common to sum all the acquired masses over the time range collected. When the chromatography is simple or all peaks are resolved, the TIC gives a good representation of the chromatography. However, when the chromatography is complex and coelutions occur, the TIC may not be a good indicator of all the peaks present in the sample because some peaks may appear as one. In this instance, one must have previous knowledge about the sample to know which components are coeluting, or resort to special deconvolution programs in order to identify the coeluting peaks.
To further complicate things, many analysts like to correlate the signals obtained by the mass spectrometer to those obtained using other detectors such as an FID. When using detectors such as an FID, it is common to do quantifications based on area percent calculations. This methodology does not translate well when using mass spectrometers as detectors because the response factors for the analytes are not the same. In addition, when coelutions occur the TIC only shows an overall profile rather than individual peaks. It may not be possible to know how much area belongs to which peak in the coelution. Time-of-Flight mass spectrometers have an advantage over scanning instruments when it comes to coelutions because in a TOFMS system the ion ratios across a peak remain constant. This is known as "spectral continuity". Because of this spectral continuity, the LECO ChromaTOF ${ }^{\circledR}$ software is capable of deconvoluting coeluting peaks and computing more precise areas for coelutions; thus area percent calculations are possible even when the TIC is not defined for coeluting peaks.

In this application note, an example is shown on the analysis of an Essential Oil imported from France called Bourdonol which is a Geranium Bourbon supplement. It will be demonstrated that accurate area percents can be calculated even for closely coeluting peaks.

## 2. Experimental Conditions

GC-Parameters:
Agilent 6890 (EPC Mode)
Column:
DB-5; $10 \mathrm{~m} \times 0.18 \mathrm{~mm} \times 0.18 \mu \mathrm{~m}$ film
Injector Temperature: $\quad 200^{\circ} \mathrm{C}$
Split Ratio: 100:1
Oven Program:
$40^{\circ} \mathrm{C}$ for 1 minute to $320^{\circ} \mathrm{C}$ at $40^{\circ} \mathrm{C} /$ minute,
hold 1 minute
Flow Rate:
Constant Flow at $1.5 \mathrm{ml} /$ minute

## MS-Parameters:

Mass Range:
Acquisition Rate:
Ion Source Temp:
Total Acquisition Time: 9 minutes

## 3. Results and Discussion

Figure 1 shows the chromatogram for the Bourdonol Essential Oil. The chromatogram was processed automatically using a signal to noise threshold of 50:1 and a peak width of 2 seconds. This means that the Processing algorithm ignores peaks that have a signal-tonoise $(S / N)$ ratio of less than $50: 1$. The Peak Find algorithm identified 150 analytes with a large number of peaks coeluting. Automatic spectral deconvolution was done and a peak table was generated followed by a library search. It is possible that more than one peak may have the same name since the Library Search algorithm cannot tell the eluting order of isomers. In addition, only the first hit for each analyte is displayed in the peak table, which may not be the correct one. Table 1 shows the peak table for this chromatogram.


Figure 1. Chromatogram of Bourdonol Essential Oil.

Another dilemma demonstrated by this sample is the calculation of area percent of coeluting peaks since some peaks coelute so closely that the software does not detect inflection points. A few examples are shown in the following figures.

Table 1. Peak Table for Bourdonol Essential Oil

| Peak\# | Name | R.T. | Quant Masses | Area | Area \% |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | Dimethyl ether | 13.44 | DT | 494,820 | 0.021 |
| 2 | Ethene, methoxy- | 14.26 | DT | 840,820 | 0.036 |
| 3 | Formic acid | 14.74 | DT | 2,264,200 | 0.097 |
| 4 | Formic acid, ethyl ester | 15.02 | DT | 138,180 | 0.006 |
| 5 | Acetic acid, anhydride with formic acid | 17.94 | DT | 515.470 | 0.022 |
| 6 | 3-Buten-2-01, 2-methyl- | 19.94 | DT | 278,430 | 0.012 |
| 7 | Propanoic acid | 28.94 | DT | 497,010 | 0.021 |
| 8 | 1-Butanol, 3-methyl- | 38.30 | DT | 559,540 | 0.024 |
| 9 | 1-Butanol, 2-methyl- | 39.24 | DT | 376,290 | 0.016 |
| 10 | 2-Butenal, 3-methyl- | 53.26 | DT | 326,550 | 0.014 |
| 11 | Pentanoic acid | 53.66 | DT | 508,680 | 0.022 |
| 12 | 1-Butanol, 3-methyl- formate | 56.24 | DT | 824,960 | 0.035 |
| 13 | 1-Pentanol, 3-methyl- | 79.68 | DT | 173,340 | 0.007 |
| 14 | 3-Hexen-1-01, (Z)- | 83.44 | DT | 3,065,200 | 0.131 |
| 15 | 2-Hexen-1-01, (E)- | 87.82 | DT | 815,640 | 0.035 |
| 16 | Acetic acid, 2-ethylbutyl ester | 99.22 | DT | 152,410 | 0.007 |
| 17 | 4-Hexen-1-ol, acetate | 106.64 | DT | 1,487,500 | 0.064 |
| 18 | Formic acid, cyclohexyl ester | 108.76 | DT | 224,110 | 0.010 |
| 19 | 3-Thujene | 108.90 | DT | 318,920 | 0.014 |
| 20 | 1R-à-Pinene | 110.92 | DT | 64,975,000 | 2.776 |
| 21 | Camphene | 115.42 | DT | 351,530 | 0.015 |
| 22 | (-)--̇-Pinene | 124.22 | DT | 3,554,100 | 0.152 |
| 23 | Bicyclo[4.1.0]heptane, 7 -(1-methylethylidene) | 126.46 | DT | 681.710 | 0.029 |
| 24 | Menth-1(8)-ene | 126.66 | DT | 1,023,400 | 0.044 |
| 25 | à-Myrcene | 129.06 | DT | 7,771,100 | 0.332 |
| 26 | 3-Heptanol, 6-methyl- | 130.22 | DT | 197,000 | 0.008 |
| 27 | à-Pinene | 132.40 | DT | 1,881,100 | 0.080 |
| 28 | à-Phellandrene | 132.56 | DT | 1,264,900 | 0.054 |
| 29 | p-Menth-8-ene, cis- | 132.76 | DT | 728,790 | 0.031 |
| 30 | 2-Hexen-1-01, acetate, (Z) | 136.06 | DT | 442,560 | 0.019 |
| 31 | 2-Menthene | 137.54 | DT | 98,416 | 0.004 |
| 32 | o-Cymene | 138.20 | DT | 5,960,500 | 0.255 |
| 33 | D-Limonene | 139.50 | DT | 4,956,600 | 0.212 |
| 34 | Menthyl acetate | 139.74 | DT | 607,600 | 0.026 |
| 35 | Cineole | 140.08 | DT | 18,371,000 | 0.785 |
| 36 | 3-Methyl-apopinene | 142.14 | DT | 129,020 | 0.006 |
| 37 | à-cis-Ocimene | 144.98 | DT | 92,548 | 0.004 |
| 38 | Benzene, buty- | 146.78 | DT | 484,350 | 0.021 |
| 39 | 4-Carene, (1S,3R,6R)-(-)- | 147.72 | DT | 108,290 | 0.005 |
| 40 | 3-Octyl methylphosphonofluoridate | 149.66 | DT | 86,085 | 0.004 |
| 41 | cis-Linalool Oxide | 151.36 | DT | 13,149,000 | 0.562 |
| 42 | trans-Linaloloxide | 155.44 | DT | 10,437,000 | 0.446 |
| 43 | 1,6-Octadien-3-ol, 3,7-dimethyl- | 159.30 | DT | 274,890,000 | 11.744 |
| 44 | Rose oxide | 161.44 | DT | 46,809,000 | 2.000 |
| 45 | Phenylethyl Alcohol | 161.70 | DT | 2,455,600 | 0.105 |
| 46 | cis-2-Pinanol | 164.12 | DT | 1,088,000 | 0.046 |
| 47 | 2H-Pyran, tetrahydro-4-methyl-2-(2-methyl-1-propenyl)- | 165.32 | DT | 15,947,000 | 0.681 |
|  | Cyclopropane, trimethyl |  |  |  |  |
| 48 | (2-methyl-1-propenylidene) | 165.56 | DT | 2,056,600 | 0.088 |
| 49 | 1,3-Cyclohexadiene, 1,2,6,6-tetramethy- | 168.54 | DT | 1,728,500 | 0.074 |
| 50 | 3,8,12-Tri-O-acetoxy-7-desoxyingol-7-one | 169.58 | DT | 193,400 | 0.008 |
| 51 | Cyclohexanol, 5-methyl-2-(1-methylethenyl). [1R-(1à,2à, 5 à $)$ ] | 169.78 | DT | 1,098,700 | 0.047 |
| 52 | Cyclohexanone, 5-methyl-2-(1-methylethyl)-, cis- | 171.98 | DT | 148,420,000 | 6.341 |
|  | Cyclohexanone, 5-methyl-2- |  |  | 148,420,000 |  |
| 53 | (1-methylethyl), trans- | 174.90 | DT | 262,220,000 | 11.202 |
| 54 | Isomenthylamine | 176.18 | DT | 760,960 | 0.033 |
| 55 | Formic acid, 2-phenylethyl ester | 176.38 | DT | 1,495,900 | 0.064 |
| 56 | 3-Cyclohexen-1-ol, 4-methyl-1-(1-methylethyl)- | 177.32 | DT | 5,975,900 | 0.255 |
| 57 | Pyrazine, 2-methoxy-3-(2-methylpropyl)- | 177.72 | DT | 4,412,000 | 0.188 |
| 58 | Cyclohexanol, 5-methyl-2-(1-methylethyl)-, (1à, 2à, 5 á)- | 178.74 | DT | 475,590 | 0.020 |
| 59 |  |  |  |  |  |
| 60 |  | 180.34 | DT | 15,324,000 | 0.655 |
| 61 | Bicyclo[3.1.0]hexane, Gisopropylidene1-methy- | 181.88 | DT | 4.592,300 | 0.196 |
|  | O-Ethyl $\mathcal{S}$-2-diethylaminoethyl |  |  | 4,592,300 |  |
| 62 | ethyiphosphonothiolate | 185.80 | DT | 1,853,000 | 0.079 |
| 63 | Tetrahydroionol | 189.74 | DT | 426,950,000 | 18.240 |
| 64 | Cyclohexane, 1methyl-3-(1methylethylidene) | 191.32 | DT | 3,167,400 | 0.135 |
| 65 | à-Citral | 191.50 | DT | 3,254,300 | 0.139 |
| 66 | 2-Octene, 2,Gdimethy- | 191.96 | DT | 3,489,100 | 0.149 |
| 67 | 3,6-Octadien-1-01, 3,7-dimethyl-, (z)- | 192.16 | DT | 3,844,900 | 0.164 |
| 68 | 1,6-Octadien-3-01, 3,7-dimethyr. (ñ)- | 195.58 | DT | 314,840,000 | 13.450 |
| 69 | 7-Hexadecyne | 196.82 | DT | 3,788,300 | 0.162 |
| 70 | trans-Citral | 197.90 | DT | 5,887,900 | 0.252 |
| 71 | 6-Octen-1-ol, 3,7-dimethyr, formate | 199.30 | DT | 242,020,000 | 10.339 |
| 72 | 2,6-Octadien-1-ol, , 3,7-dimethyr, acetate, (z) $=$ | 200.14 | DT | 12,538,000 | 0.536 |
| 73 | 1,6-Octadien-3-0, 3, 3,-dimethyr, formate | 204.70 | DT | 113,070,000 | 4.831 |
| 74 | 6-Octenoic acid, 3,7dimethyt, methyl ester | 207.20 | DT | 9,147,000 | 0.391 |
| 75 | 4-Chloro-3-n-hexyltetrahydropyrane | 208.02 | DT | 45,109 | 0.002 |
| 76 | Neric acid | 209.64 | DT | 4,046,200 | 0.173 |
| 77 | 2-Butenoic acid, 2methyl, 2-methylpropyl ester, (E) | 209.90 | DT | 2,936,300 | 0.125 |
| 78 | 1,2-Cyclohexanediol, 3methyl-6-(1-methylethyl), (1à, 2á, 3á, 6 à) | 210.56 | DT | 740,620 | 0.032 |
| 79 | Cyclohexanol, 2methyt-3-(1-methylethenyl) acetate, (1à,2à,3à) | 210.66 | DT | 1,075,000 | 0.046 |
| 80 | Z-11,12-Epoxytetradecar1-ol | 211.22 | DT | 253,040 | 0.011 |
| 81 | cis-2,6-Dimethy-2,6-octadiene | 214.44 | DT | 12,643,000 | 0.540 |
| 82 | à-Cubebene | 215.00 | DT | 1,063,200 | 0.045 |
| 83 | Phenol, 2-methoxy-3-2-propenyl)- | 215.66 | DT | 1,866,500 | 0.080 |
| 84 | Neric acid | 215.96 | DT | 6,547,800 | 0.280 |
| 85 | Nerol acetate | 216.64 | DT | 2,759,800 | 0.118 |
| 86 | 2,6-Octadien-1-ol, 3,7-dimethyr, acetate, (Z) | 220.38 | DT | 5,651,600 | 0.241 |
| 87 | Copaene | 220.56 | DT | 2,173,000 | 0.093 |
| 88 | 3,9-Epoxypregn-16-ene-14-18-diol-20-one, 7,11-diacetoxy-3-methoxy- | 221.44 | DT | 102,440 | 0.004 |
| 89 | Cyclohexene, Getheny-6-methyl-1-(1-methylethyl)3-(1-methylethylidene), (S)- | 222.08 | DT | 6,905,100 | 0.295 |
| 90 | Cyclobuta[1,2:3,4]dicyclopenten, decahydro3a-methy-6-methylene 1 -(1-methylethyl), [1S-(1à,3aà,3bá,6aá,6bà)] | 222.50 | DT | 2,793,500 | 0.119 |


| Peak \# | Name | R.T. | Quant Masses | Area | Area \% |
| :---: | :---: | :---: | :---: | :---: | :---: |
| 91 | $\begin{aligned} & \text { [5-(3-Methoxymethoxy10,13-dimethy- } \\ & \text { 2,3,4,9,10,11,12,13,14,15,16,17 } \\ & \text { dodechydro1H-cyclopenta[a]phenanthrea } \\ & \text { 17-yl)-hex-1-ynyl]-frime } \end{aligned}$ | 223.16 | DT | 32,428 | 0.001 |
| 92 | Cyclohexane, retheny-1-methyl-2,4-bis (1-methylethenyl) | 223.44 | DT | 2,041,300 | 0.087 |
| 93 | Butane, 1,1-sulfinylbis- | 223.58 | DT | 1,127,600 | 0.048 |
| 94 | Diphenyl ether | 224.76 | DT | 35,224,000 | 1.505 |
| 95 | 2-Methyl-1-undecanol | 225.60 | DT | 78,978 | 0.003 |
| 96 | 1,4-Methanoazulene, decahydrө4,8,8- trimethyl-9-methylene, [1S-(1à,3aá, 4à, 8aá)] | 226.88 | DT | 601,500 | 0.026 |
| 97 | Isoledene | 228.00 | DT | 2,731,100 | 0.117 |
| 98 | Caryophyllene | 229.40 | DT | 26,807,000 | 1.145 |
| 99 | 6-Octen-1-ol, 3,7-dimethyr, propanoate | 230.96 | DT | 520,370 | 0.022 |
| 100 | á-Cubebene | 231.02 | DT | 196,010 | 0.008 |
| 101 | 2,4,6-Octatriene, 2,6dimethyr, (E,Z)- | 231.26 | DT | 22,424 | 0.001 |
| 102 | Citronellyl propionate | 231.90 | DT | 7,112,000 | 0.304 |
| 103 | à-Guaiene | 232.48 | DT | 9,834,000 | 0.420 |
| 104 | Isoledene | 233.44 | DT | 7,680,600 | 0.328 |
| 105 | Butanoic acid, 3-methyl-, 1-etheny-1,5- dimethy-4-hexenyl ester | 233.82 | DT | 1,327,100 | 0.057 |
| 106 | Isoledene | 234.80 | DT | 4,434,000 | 0.189 |
| 107 | à-Caryophyllene | 235.82 | DT | 2,852,200 | 0.122 |
| 108 | (+)-Cyclosativene | 237.30 | DT | 2,349,200 | 0.100 |
| 109 | Geranyl propionate | 237.72 | DT | 17,622,000 | 0.753 |
| 110 | (-)-Calamenene | 238.44 | DT | 1,280,600 | 0.055 |
| 111 | 2-Isopropeny-4a,8-dimethy-1,2,3,4,4a,5, 6,7-octahydronaphthalene | 239.58 | DT | 3,110,700 | 0.133 |
| 112 | Germacrene D | 240.78 | DT | 626,070 | 0.027 |
| 113 | Isolongifolanr-ol | 241.58 | DT | 3,260,100 | 0.139 |
| 114 | (ñ)-Cadinene | 242.48 | DT | 7,077,300 | 0.302 |
| 115 | 3,7-Cyclodecadien1-one, 10-(1-methylethenyl), (E,E)- | 243.46 | DT | 852,410 | 0.036 |
| 116 | Azulene, 1,2,3,4,5,6,7,8octahydro-1, 4-dimethyl7-(1-methylethenyl), [1S(1à,4à,7à)) | 243.94 | DT | 1,620,200 | 0.069 |
| 117 | Isoledene | 244.12 | DT | 1,066,100 | 0.046 |
| 118 | à-Bulnesene | 245.22 | DT | 11,539,000 | 0.493 |
| 119 | Azulene, 1,2,3,4,5,6,7,8octahydro-1,4-dimethyl7-(1-methylethylidene), ( 1 S -cis)- | 245.70 | DT | 615,200 | 0.026 |
| 120 | 2-Methy-2,3-pentanediol | 246.26 | DT | 257,550 | 0.011 |
| 121 | ç-Cadinene | 246.78 | DT | 170,350 | 0.007 |
| 122 | à-Guaiene | 247.04 | DT | 504,040 | 0.022 |
| 123 | Phenol, 2-methoxy-4-(2-propenyl), acetate | 247.40 | DT | 6,365,200 | 0.272 |
| 124 | Butanoic acid, 3,7dimethyt6-octenyl ester | 247.66 | DT | 8,572,800 | 0.366 |
| 125 | à-Cadinene, (t)- | 248.16 | DT | 2,065,400 | 0.088 |
| 126 | Propanoic acid, 2methyl-, 3,7-dimethyt2, 6 -octadienyl ester, (E) | 249.30 | DT | 384,520 | 0.016 |
| 127 | 5-(4,4,3-Trimethylpentyl)2-pyridinecarboxylic acid | 250.28 | DT | 24,983 | 0.001 |
| 128 | 1H-Cycloprop[e]azulene, 1a,2,34,4a,5,6,7b-octahydro-1,1,4,7-tetramethyt, [1aR(1aà,4à,4aá,7bà)ł | 250.84 | DT | 1,287,500 | 0.055 |
| 129 | Acetic acid, cyano, methyl ester | 251.58 | DT | 36,356 | 0.002 |
| 130 | 6-(3-Isopropenylcycloprop1-enyl)-6-methylhept 3-en-2-one | 251.88 | DT | 207,890 | 0.009 |
| 131 | Eudesma3,7(11)-diene | 252.00 | DT | 399,980 | 0.017 |
| 132 | Nerol acetate | 253.36 | DT | 7,039,700 | 0.301 |
| 133 | Guanidine, monothiocyanate | 254.56 | DT | 65,114 | 0.003 |
| 134 | à-Guaiene | 255.80 | DT | 5,723,800 | 0.245 |
| 135 | 3,4-Hexanedione, 2,2,5trimethy- | 256.72 | DT | 122,800 | 0.005 |
| 136 | 2-Butenoic acid, 2-methyl, 2-phenylethyl ester, (E)- | 258.34 | DT | 732,570 | 0.031 |
| 137 | 4-(3,3-Dimethy-but-1-ynyl)-4-hydroxy-2,6, 6-trimethylcyclohe 2 -enone | 259.32 | DT | 340,100 | 0.015 |
| 138 | Caryophyllene oxide | 259.52 | DT | 1,346,400 | 0.058 |
| 139 | Benzophenone | 266.14 | DT | 1,116,800 | 0.048 |
| 140 | cis-3-Hexenyl phenyl acetate | 266.44 | DT | 522,230 | 0.022 |
| 141 | Citronelly liglate | 269.90 | DT | 123,000 | 0.005 |
| 142 | Bicyclo[[3.1.1]heptane, 2,6,6itrimethy- | 271.70 | DT | 1,348,300 | 0.058 |
| 143 | Geranyl tiglate | 277.50 | DT | 507,260 | 0.022 |
| 144 | cis-Linalool Oxide | 285.04 | DT | 114,300 | 0.005 |
| 145 | (-)-cis-Carane | 320.44 | DT | 1,136,900 | 0.049 |
| 146 | Nery phenylacetate | 326.10 | DT | 2,857,800 | 0.122 |
| 147 | 6-Octen-1-ol, 3,7-dimethyr, acetate | 329.16 | DT | 1,512,800 | 0.065 |
| 148 | 2,6,10-Dodecatrien-1-01, 3,7,11-trimethyt, (E,E)- | 334.04 | DT | 933,200 | 0.040 |
| 149 | 2-Naphthuric acid | 472.38 | DT | 1,024,900 | 0.044 |
| 150 | Capric acid triglyceride | 490.00 | DT | 1,024,400 | 0.044 |
|  | Total |  |  | 2,340,800,000 | 100.000 |

The following examples (from the sample above) illustrate the problems associated with area percent calculations using a mass spectrometer. In the first example, the peaks are calculated using the natural inflection points and valleys in the TIC as detected by most software packages. Normally a perpendicular is drawn between valleys of coeluting compounds, but this is only an approximation. The problem is compounded when peaks coelute so closely that there are no defined valleys between them. In this instance, most software packages can only guess at the location to draw perpendiculars from the peak apexes.

Figure 2 shows the TIC section of the Bourdonol Essential Oil between 213.5 and 217.5 seconds. Inspection of the TIC only reveals the presence of four peaks (in actuality there are five peaks). It is difficult to assign integration limits when the peaks are not well defined or there are no defined valleys between the peaks. This is the main reason why long chromatographic runs are needed by most chromatographic platforms to ensure accurate peak areas. Table 2 shows a section of the peak table as calculated using the area percent from the TIC. These numbers are incorrect for the reasons noted above.


Figure 2. Chromatogram Section of Bourdonol Essential Oil (Coeluting Peaks).

Table 2. Erroneous Area Percent Calculation Based on the TIC

| Peak \# | Name | R.T. | Quant <br> Masses | AREA | AREA \% |
| :---: | :--- | :---: | :---: | :---: | :---: |
| 81 | cis-2,6-Dimethy+2,6-octadiene | 214.44 | TIC | 14422000 | 0.549 |
| 82 | Phenol, 2-methoxy-3-(2- <br> propenyl) | 215.66 | TIC | 4357900 | 0.166 |
| 83 | Neric acid | 215.96 | TIC | 2800200 | 0.106 |
| 84 | 2,6-Octadien-1-ol, 3,7-dimethyt, <br> acetate, (Z) | 216.64 | TIC | 3069800 | 0.117 |

The areas calculated from the TIC are incorrect because the software cannot assign good integration limits. First, the area for peak 81 (cis-2,6-Dimethyl-2,6-octadiene) is only approximate because there is not enough valley between the adjacent peak. Second, the area for peak 82 (2-methoxy-3-(2-propenyl)-phenol is also not correct because it also has added contribution from peak 83 (Neric acid). The best the software can do is to draw a perpendicular between peaks 82 and 83 . Peak 83 has the same problem as peak 82 where the coelution is so close that good integration areas are difficult to calculate. The area for peak 84 is underestimated because there is not enough valley between the adjacent peak.

The ChromaTOF software can actually calculate accurate areas for each of the peaks shown in the example based on the spectral continuity of the peaks. By specifying in the QUANT MASSES section of the peak table a deconvoluted TIC (DTIC) the software is instructed to calculate how much contribution each peak makes to the TIC so that good integration limits can be calculated. Table 3 shows the area percent calculation based on the deconvoluted TIC. Note that the ChromaTOF software detects the presence of five peaks. Peak 82 (Cubebene) is not obvious from the TIC. Now the integration limits are set correctly and areas that are more accurate can be calculated for more reliable area percent results.

Table 3. Area Percent Calculation Based on the DTIC as Computed by Software

| Peak \# | Name | R.T. | Quant <br> Masses | Area | Area \% |
| :---: | :--- | :---: | :---: | :---: | :---: |
| 81 | cis-2,6-Dimethy-2,6-octadiene | $\$ 214.44$ | DT | $12,150,000$ | 0.516 |
| 82 | à-Cubebene | $\$ 215.00$ | DT | $1,058,700$ | 0.045 |
| 83 | Phenol, 2-methoxy-3-(2- <br> propenyl) | $\$ 215.66$ | DT | $1,860,200$ | 0.079 |
| 84 | Neric acid | $\$ 215.96$ | DT | $6,133,700$ | 0.260 |
| 85 | 2,6-Octadien-1-ol, 3,7-dimethyr, <br> acetate, (Z) | $\$ 216.64$ | DT | $2,034,200$ | 0.086 |



Figure 3. Chromatogram Section of Bourdonol Essential Oil (Coeluting Peaks with DTIC).

Figure 3 shows the same chromatographic section as Figure 2 with the deconvoluted TIC sections for each peak. The software can easily calculate integration limits without trying to fit gaussian shapes in the coelution (to calculate areas) as some other deconvolution packages do. The areas are calculated on the actual shape of the peak. Note that the Neric acid peak is fronting and does not have a gaussian shape, but the software gives a good area percent calculation.

A more extreme case is shown in Figure 4 for a pair of peaks that coelute very closely in the region of 160.5 and 162.5 seconds. In addition to the close coelution, the concentration difference between the peaks is substantial. In this case, most chromatographic platforms would not detect a coelution since the TIC shows a well-defined peak. Using the TIC as the quantification mass, the software calculates the areas for both peaks the same since it cannot define inflection points or valleys for close coeluting peaks. Table 4 shows the area percent calculation using the TIC while Table 5 shows the peak areas based on the DTIC.


Figure 4. Chromatogram Section of Bourdonol Essential Oil (Closely Coeluting Peaks).

Table 4. Area Percent Calculation Based on the TIC as Computed by Software for Above Peaks

| Peak\# | Name | R.T. | Quant <br> Masses | Area | Area \% |
| :---: | :--- | :---: | :---: | :---: | :---: |
| 44 | Rose oxide | 161.44 | T | $49,124,000$ | 2.0366 |
| 45 | Phenylethyl Alcohol | 161.70 | T | $49,138,000$ | 2.0372 |

Table 5. Area Percent Calculation Based on the DTIC as Computed by Software for Above Peaks

| Peak \# | Name | R.T. | Quant <br> Masses | Area | Area \% |
| :---: | :--- | :---: | :---: | :---: | :---: |
| 44 | Rose oxide | 161.44 | DT | $46,809,000$ | 1.981 |
| 45 | Phenylethyl Alcohol | 161.70 | DT | $2,455,600$ | 0.104 |

## 4. Conclusions

The analysis of Bourdonol Essential Oil demonstrates that accurate area percent calculations can be easily done even when chromatographic coelutions take place. By instructing the processing software, it is possible to obtain accurate area contributions from coelutions even when peaks have a non-gaussian shape. This allows the chromatography to be compressed in time so that more samples can be analyzed.

The strength of the Pegasus GC-TOFMS for the analysis of these complex mixtures lies in its automated data handling capabilities. Peak finding and spectral determination is possible due to the uniqueness of the data acquisition system, which allows up to 500 spectra/ second to be collected while simultaneously obtaining non-skewed spectra (spectral continuity). Library searching and area percent determinations can be accomplished very rapidly, improving analytical results and productivity.


