

Thermo Fisher SCIENTIFIC

The Doctor Did Not Prescribe Irgafos:

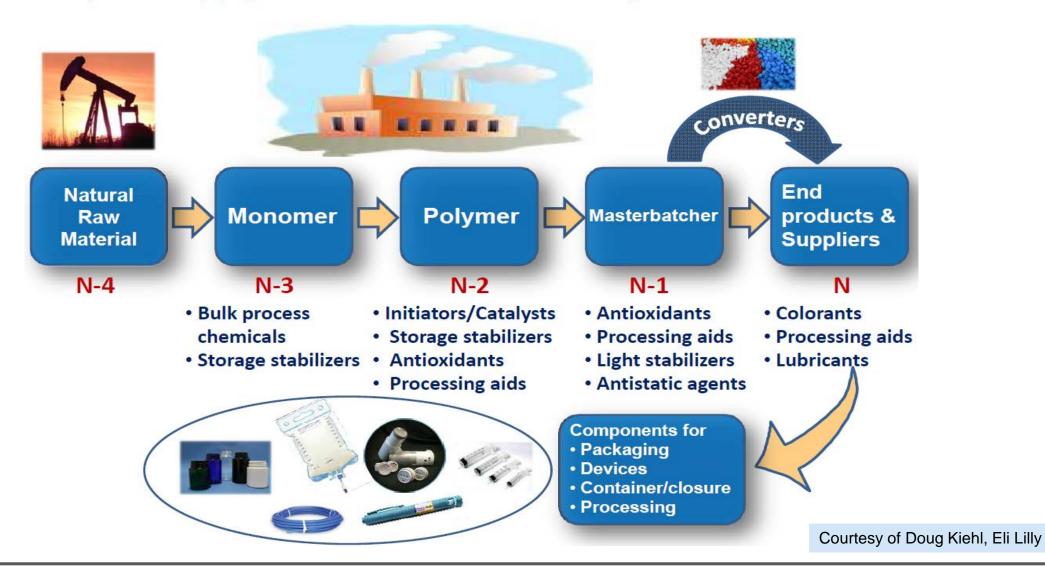
A Solution for Extractables and Leachables Analysis

Outline

- Introduction
 - Why conduct Extractables & Leachable (E&L) analysis?
 - Industries need Extractable & Leachable analysis
 - Regulations and guidelines from agencies and industry consortiums
- Thermo Scientific™ solutions for extractable & leachable analysis
 - Multiple instrumentations and technologies
 - ASE, CAD
 - GCMS
 - LCMS
 - ICP-MS
 - Data analysis software and database/spectral library

Plastics – A Complex Process

Polymer Supply Chain for Pharmaceutically Relevant Materials



Definitions

- Extractable (E)
 - Compounds that migrate from the surface under more aggressive conditions of exposure (solvent, time, and temperature). <u>Controlled extraction study</u>
- Leachable (L)

Degradation Products

Solvent-material Interactions

 Compounds that migrate from the contact surface to drug formulation under normal conditions of exposure. <u>Bioproduction, formulated drug, or device simulants</u>

Additives Impurities Polymer Components Extractables Leachables

Product Quality

Known extractables

Extractable – drug formulation interactions

Extractable modified by process conditions

Some leachables are subset of extractables, but this is not always the case.

Safety/risk assessment → Regulatory submission → Market

The Broad Market of Extractable & Leachable Analysis









Single-use systems for bioproduction and storage











Medicine container/packaging, implant and diagnostic devices



Food packaging



Printing ink and adhesives



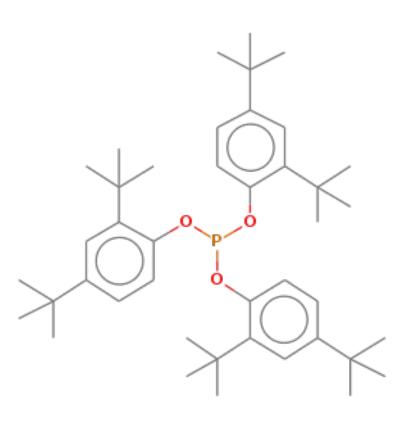
Wearable consumer electronic products



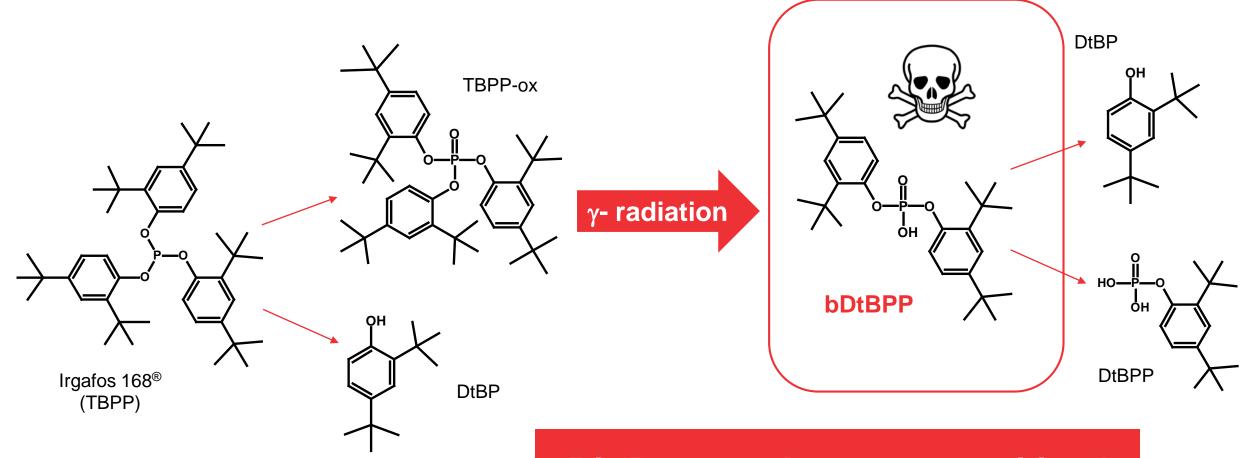
Trade name - Irgafos® 168 (BASF) tris(2,4-di-tert.-butylphenyl)phosphite.

- Polymer additive stabilizer
- Protects plastic from thermooxidative degradation
- Improves the strength and durability
- Used in a range of plastic films

including Single Use BioProcess films



Irgafos 168 is safe – until you irradiate it



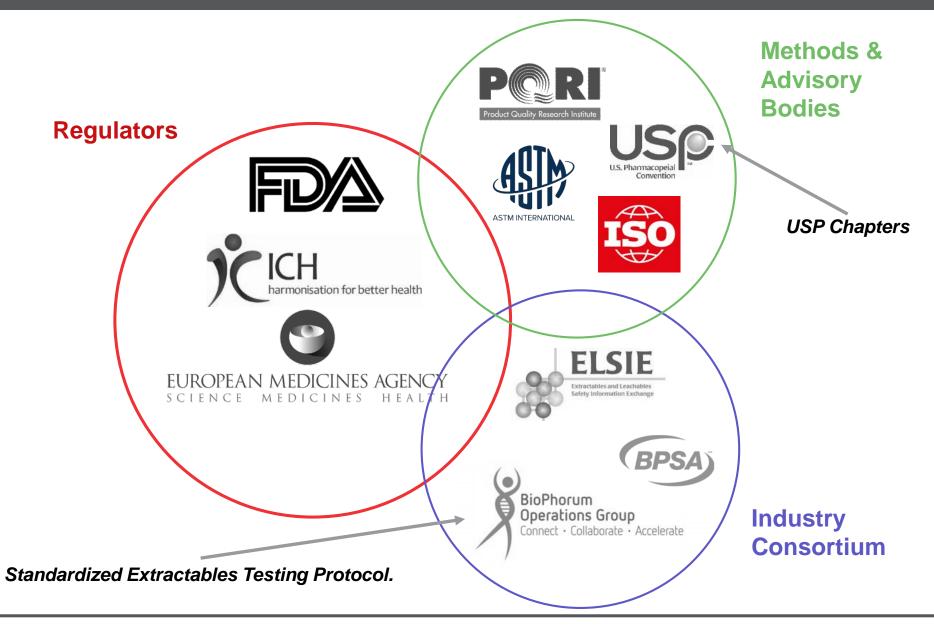
BioProcess bags are sterilized with γ -radiation

¹ Fouyer K et al (2012) Anal. Chem. 84 (20): 8642-8649.

The Importance and Challenges of E&L Analyses

- Safety
 - Extractables and Leachables are chemical impurities- they could contaminate medicine, food, and drink, affect product quality, and cause adverse effects for consumers.
- Regulations
 - E&L analyses are highly regulated and required for marketing approval of new products.
 - More regulations have been promulgated along the way and will continue into the future.
- E&L analysis is challenging and complex
 - It requires multiple instruments, expertise, data processing software, and database.
 - It needs both identification and quantitation. Unknown structure elucidation is challenging.
 - It must start at early stage to be ready for product launch.

E&L Regulatory and Method landscape



Industry Group: BioPhorum Operations Group (BPOG) The Global Community

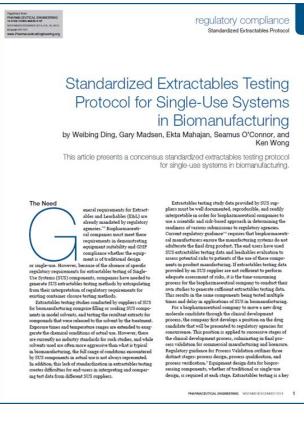
http://www.biophorum.com/





BPOG members are leading biopharmaceutical companies around the world to create an environment where the global biopharmaceutical industry can collaborate and accelerate their rate of progress, for the benefit of all.

Many Technical Resources on their website http://www.biophorum.com/



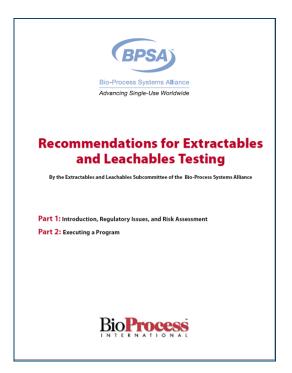
Industry Group: Bio-Process Systems Alliance (BPSA) - Published Technical Guides



BPSA members include industry-leasing manufacturers of singleuse biological processing products. BPSA is encouraging and accelerating the adoption of technologies used in the production of biopharmaceuticals and vaccines.

Many Technical Guides on their website bpsalliance.org/







ISO 10993

Biological Evaluation of Medical Devices

Sample preparation and Extraction

- Part 12: Sample preparation and reference materials
 Material Characterization
- Part 18: Chemical characterization of materials
 - GCMS SVOC
 - LCMS NVOC
 - ICP- MS Elemental impurities
 - Other analytical methods FTIR...

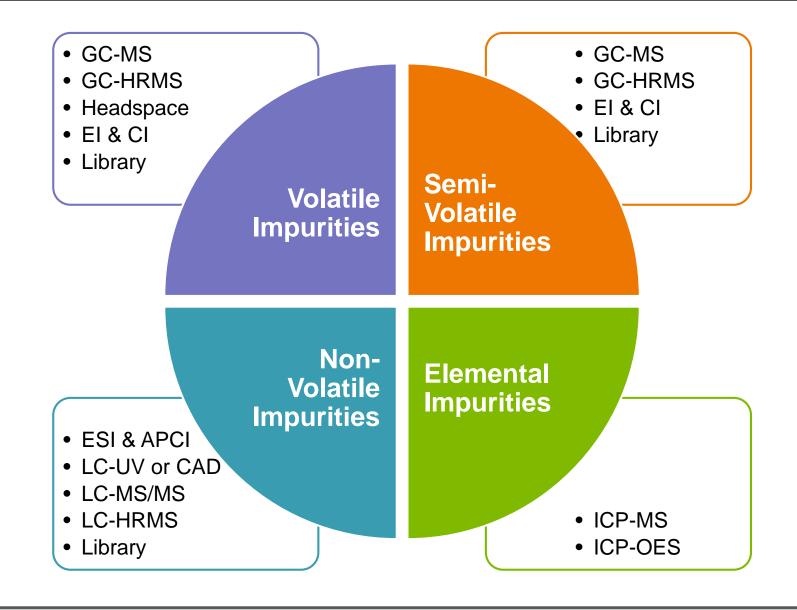


ASTM D7210

Standard Practice for Extraction of Additive in Polyolefin Plastics

- USP <1663> lists ASE as a possible option to perform extractable studies
- ASTM D7210 also lists ASE as one of the options to extract antioxidants from polymeric matrices.

Analysis of Extractables & Leachables



Thermo Scientific Instruments, Software, Consumables for Impurity ID & E&L Analysis





ICP-MS



Elemental Impurity – Thermo Scientific iCAP RQ ICP-MS and Qtegra Software

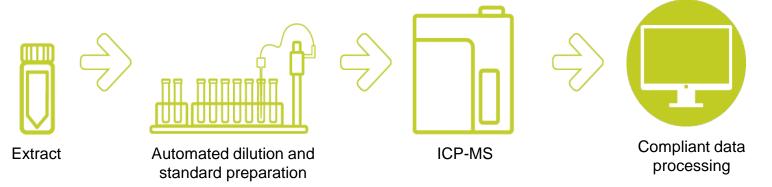
Thermo Scientific™ iCAP™ RQ ICP-MS



Thermo Scientific™ Qtegra™ ISDS software.

Integrated CFR compliance tools kit Full USP 233 & ICH Q3D method capabilities

Compliance with USP <232>/<233>



Using an in-line autodilution and autocalibration system significantly lowers the risk of human error and contamination of the samples, standards, or blanks.



Thermo Scientific™ iCAP™ 7000 Plus ICP-OES

Principle and Best Practices Recommended

Extraction efficiency

Accelerated solvent extraction technique (30 minutes)

Guidelines list ASE as an option for extractable studies:

- USP Chapter <1663> & <1664>
- Product Quality Research Institute (PQRI)
- BioPhorum Operations Group (BPOG)
- ASTM D7210



Soxhlet (24 Hours)



Heated agitation (2-30 days)

Advantage

- Automated with intelligent solvent management system
- Reduce extraction times
- Reduce solvent consumption
- Increase extraction efficiency
- Nitrogen flush gas prevents the oxidation of the extractables
- Working very well for wide range of polymers, especially for ultra high molecular weight cross linked polyethylene -Dr. Vas

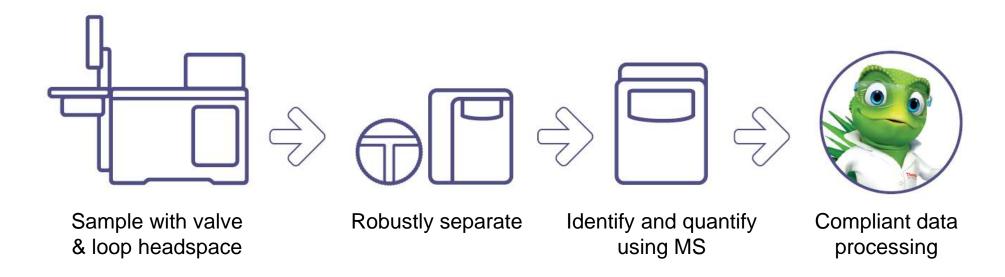
Time

Comparison of Soxhlet vs. Thermo Scientific Dionex ASE 350 System

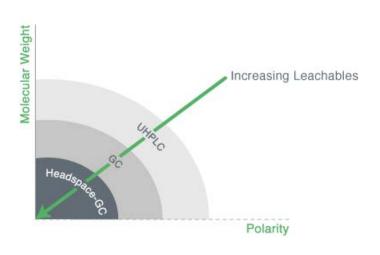
Parameter	Soxhlet	Thermo Scientific™ Dionex™ ASE™ 350 System
Extract solvent used per sample (mL)	160	<30
Total extraction time per sample (min)	1440	<30
Extracted compounds	Same	Same
Extracts peak Intensity Ratio Accelerated Solvent Extraction/Soxhlet		1.4x to 90x

Dionex ASE 350 system delivers faster extractions using less solvent

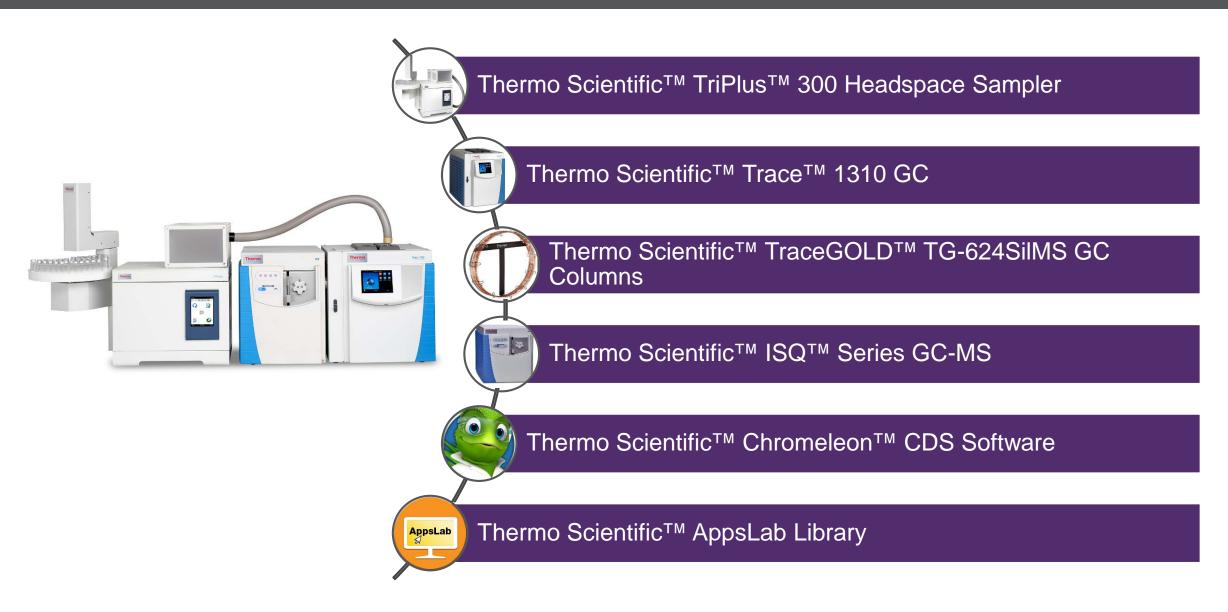
Volatile Impurities Workflow



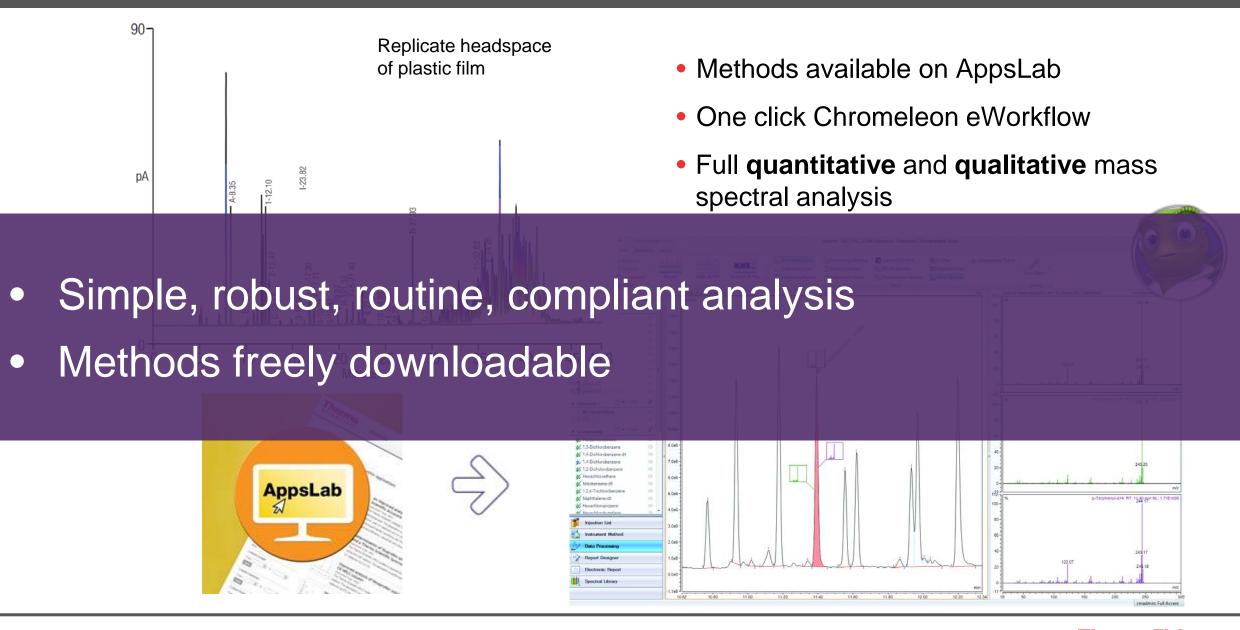
- Analogous to USP 467 Residual solvents workflow
- Molecules are generally known, or simple to identify
- Routine compliant quantification



Complete Technologies for Volatile Impurities



Volatiles Headspace Analysis



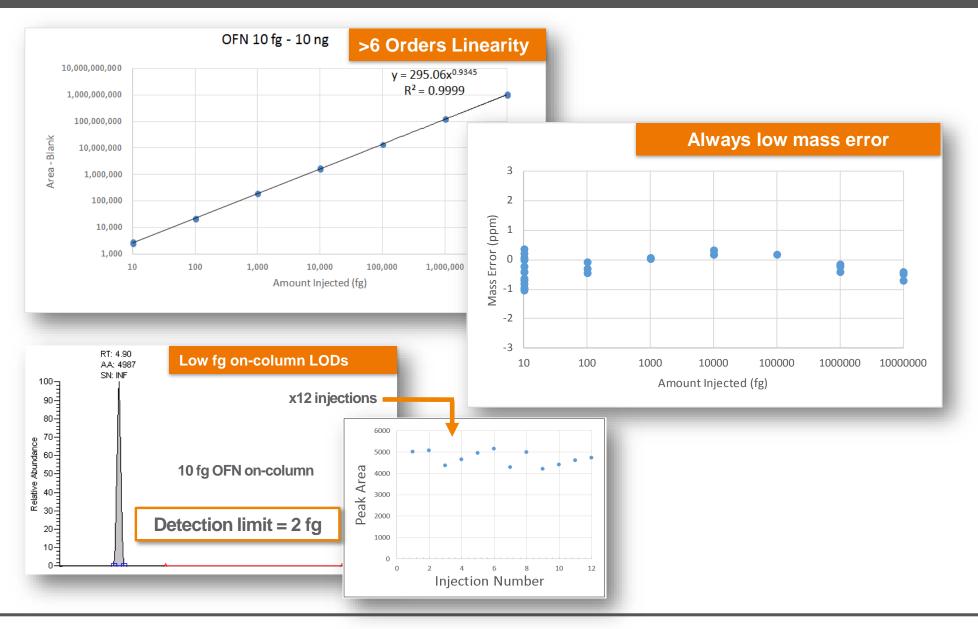
High Resolution GCMS for Impurity Identification and Quantification Workflow



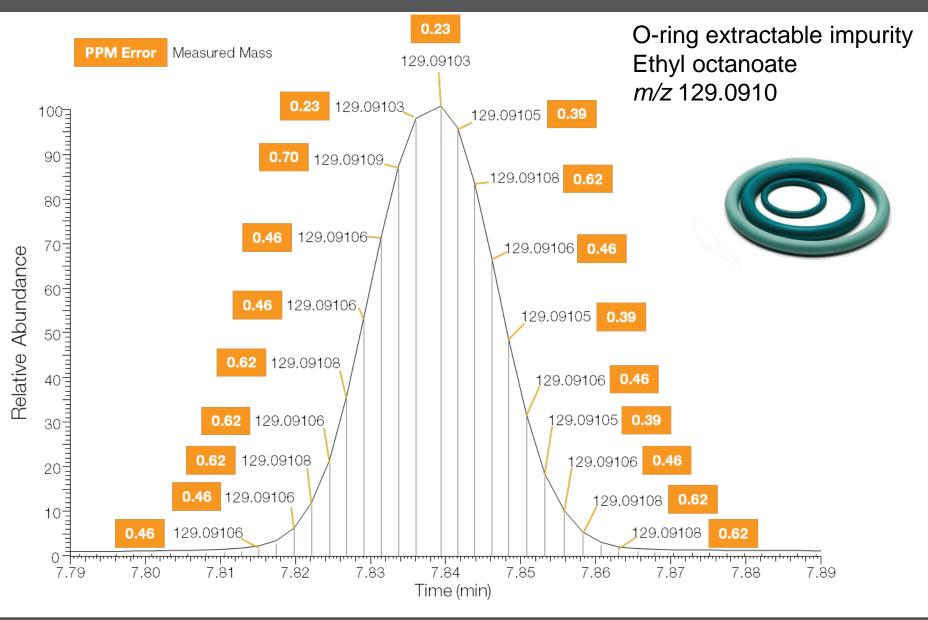
- Deconvolve, identify and quantify even the narrowest GC peaks
- Unambiguously calculate empirical formulae
- No need to average scans
- Simplify data review and report
 - Thermo Scientific™ Q Exactive™ GC Orbitrap™ GC-MS/MS system
 - Thermo Scientific[™] TraceFinder[™] software



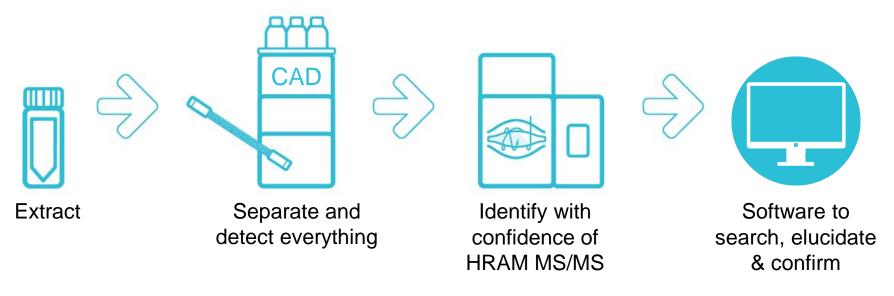
Detect, Quantify and Identify at Any Concentration



Scan Speed and Accurate Mass Error Across a Peak



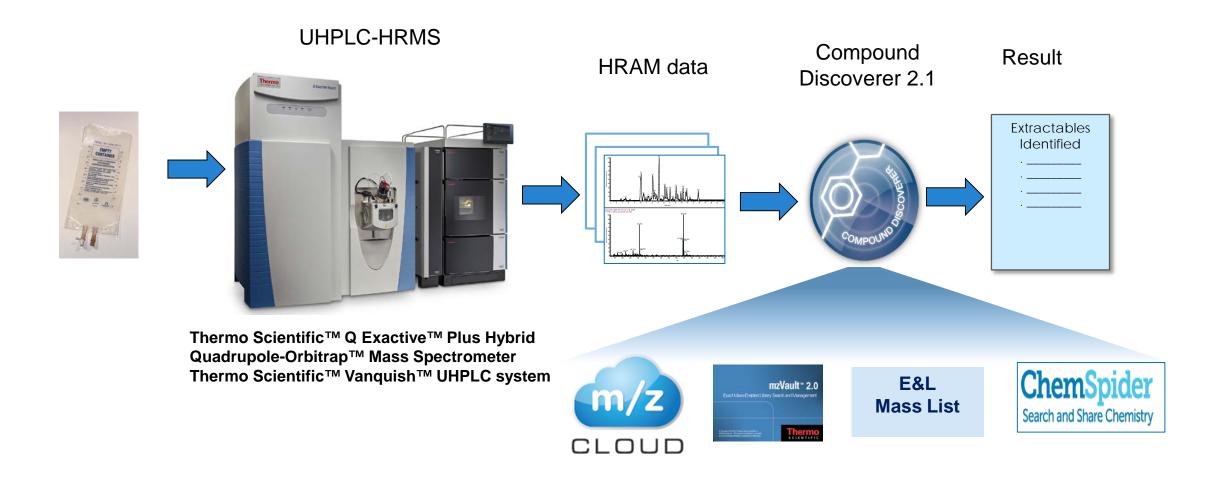
Non-Volatiles Unknown Identification Workflow



- Catch everything with orthogonal detection
- Detect in both positive and negative ion mode
- Have absolute confidence in elemental composition
- Get full sub-structural information
- Search wide variety of data sources in parallel

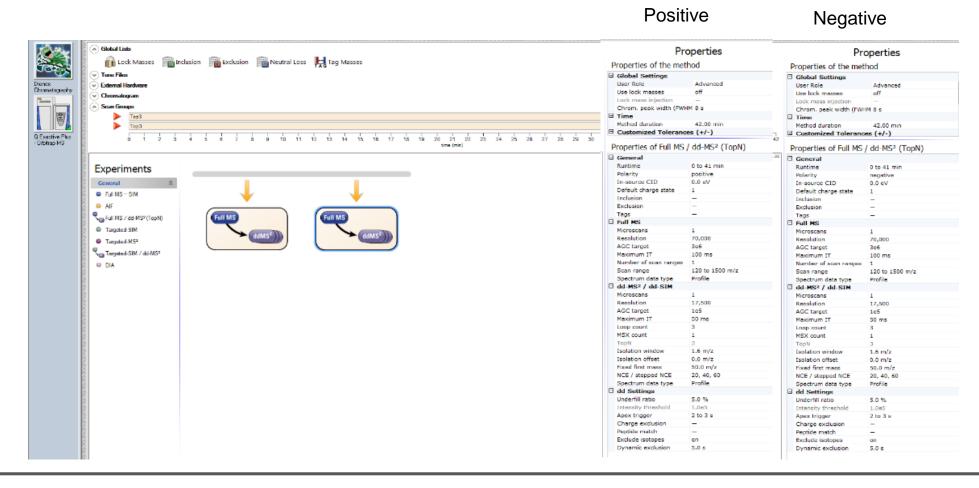


LC-HRMS Analysis for IV Bag Extractables

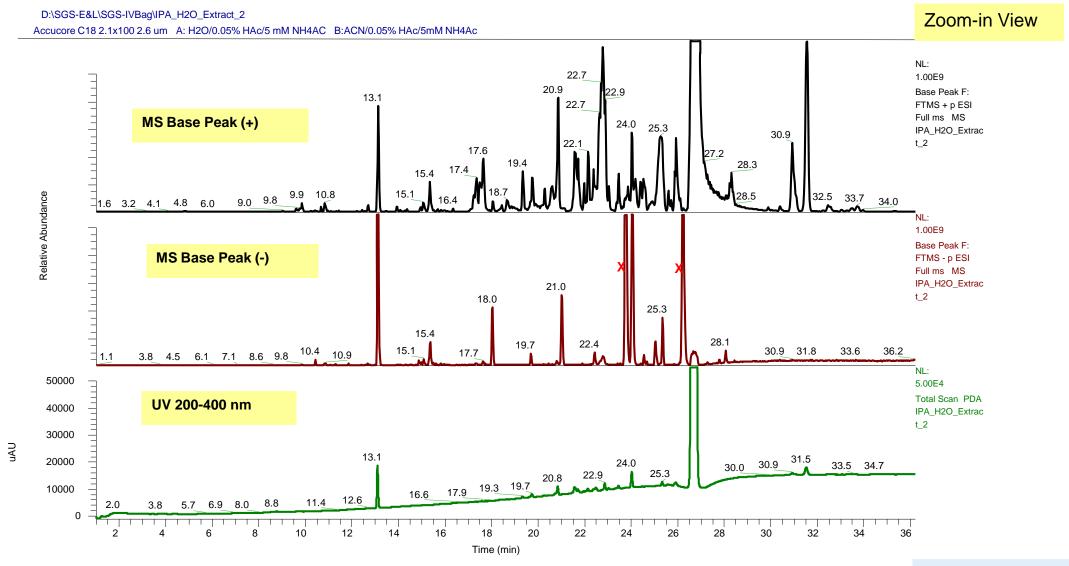


MS Method: High Resolution Accurate Mass Untargeted Screening

- HRAM untargeted screening with polarity switching
- ESI/APCI full scan MS and data-dependent top 3 MS/MS data with polarity switching using 70K &17.5K resolution for FMS and HCD MS² respectively. Stepped NCE: 30, 45, 60.



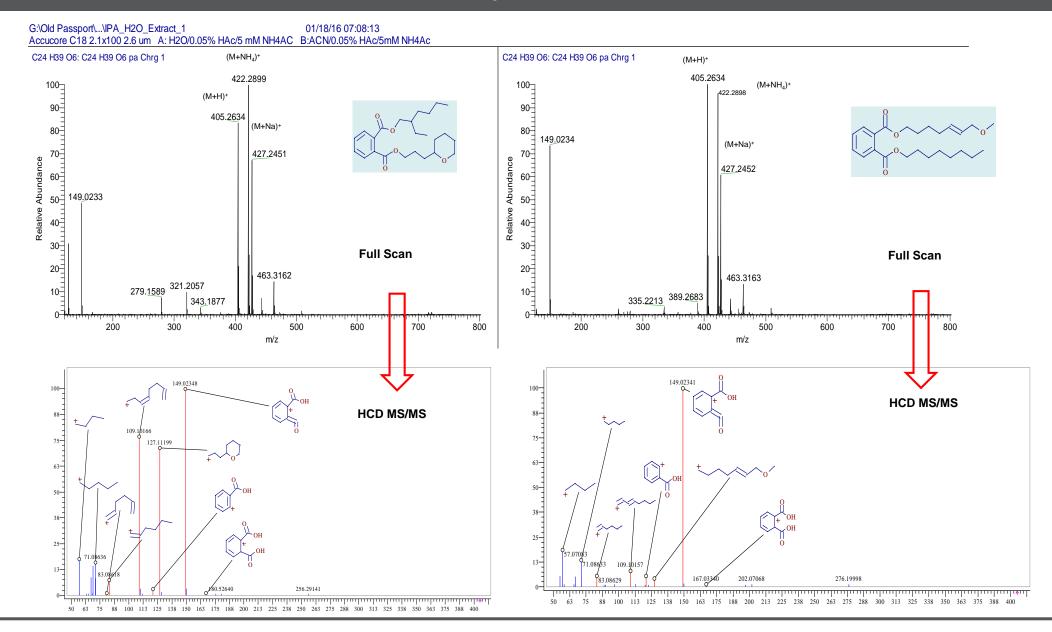
IV Bag IPA-H2O Extract - Positive Negative Full Scan, MSMS and UV in One Run



X = present in the blank

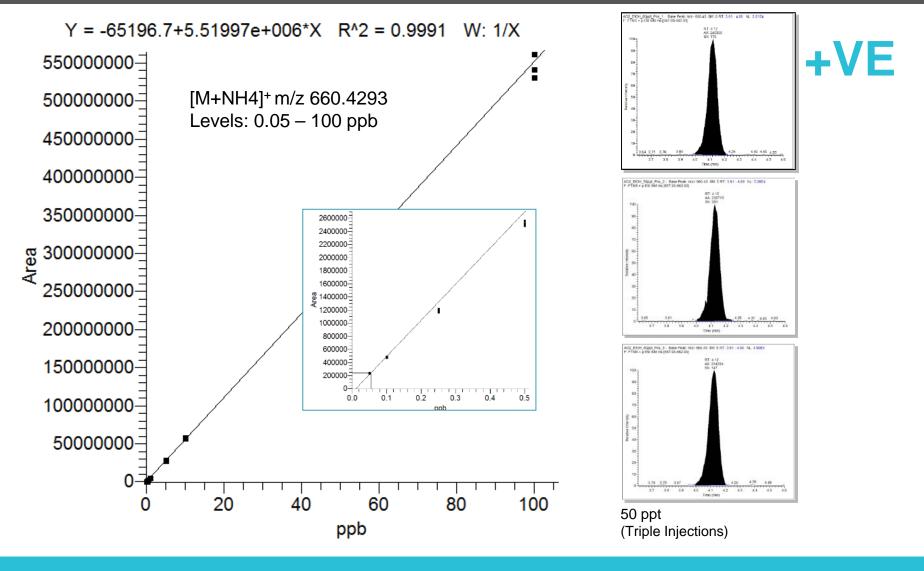


Full Scan and HCD MS/MS Data for Component ID and Structure Elucidation





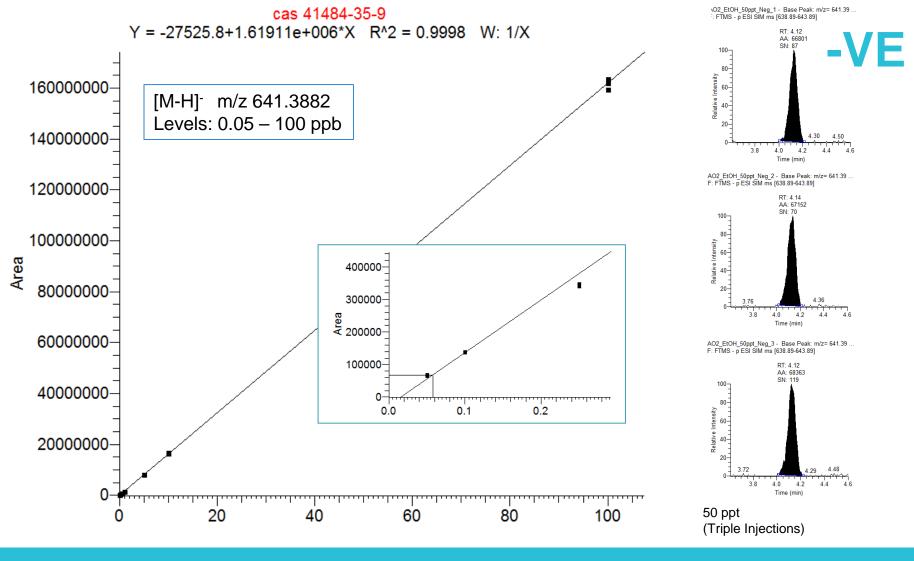
Quantify Non-Volatile Extractables – Irganox1035



Detect in positive and negative ion mode in the same run



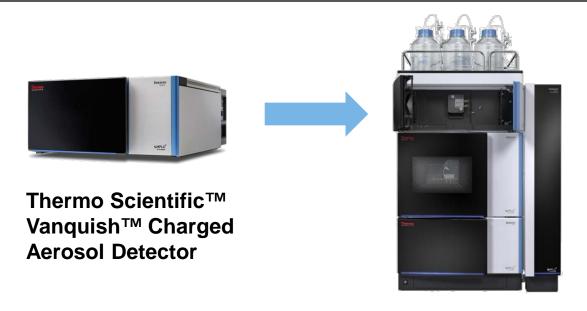
Quantify Non-Volatile Extractables – Example Irganox1035



Full sensitivity in negative ion mode

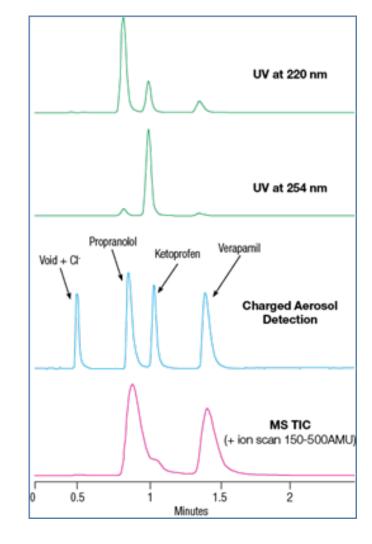


See What You're Missing with Charged Aerosol Detection

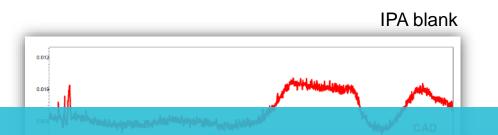


- Detect components without chromophore
- Quantify without exact standards
 - Relative quantification due to consistent response
 - Use virtually any standard for simplified AET calculations
- Consistent analyte response
- Four orders dynamic range
- USP <1663> listed

Comparison of Charged Aerosol Detection Result to UV and MS

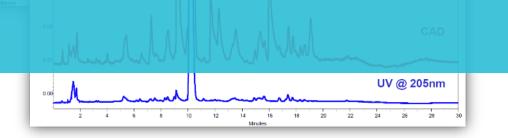


CAD for Extractables and Leachables



"Virtually universal detection"

- "able to detect and measure a large number of compounds that were completely transparent to UV detection."
- "CAD can provide a level of information that has until now been lacking with methods such as UV and even Mass Spec."



Data from ESA Biosciences, Inc., Chelmsford, MA

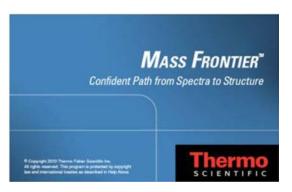
Data Analysis Software, Database and Spectral Library for E&L Analysis



Thermo Scientific[™] Compound Discoverer[™] 2.1



Thermo Scientific™ TraceFinder™
Targeted Screening and Quan



Thermo Scientific™ Mass Frontier™ Spectral Interpretation software



mzCloud is a trademark of HighChem LLC, Slovakia



mzVault™ 2.0 Library Search and Manager



E&L Compound Database

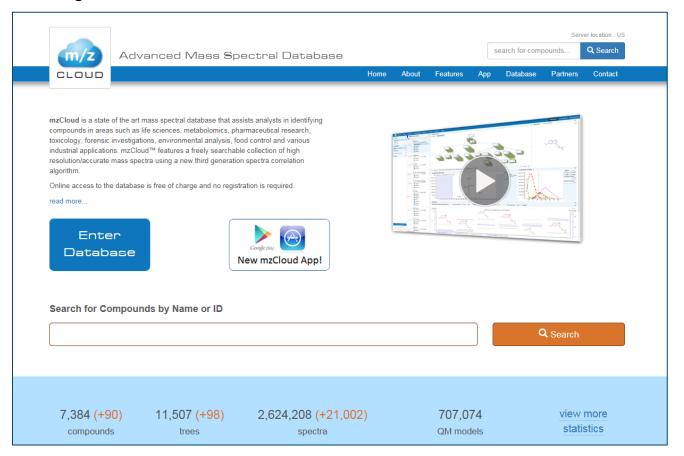


mzCloud[™] - HR Mass Spectral Database



mzCloud™ Database: Free and cloud-based (www.mzcloud.org)

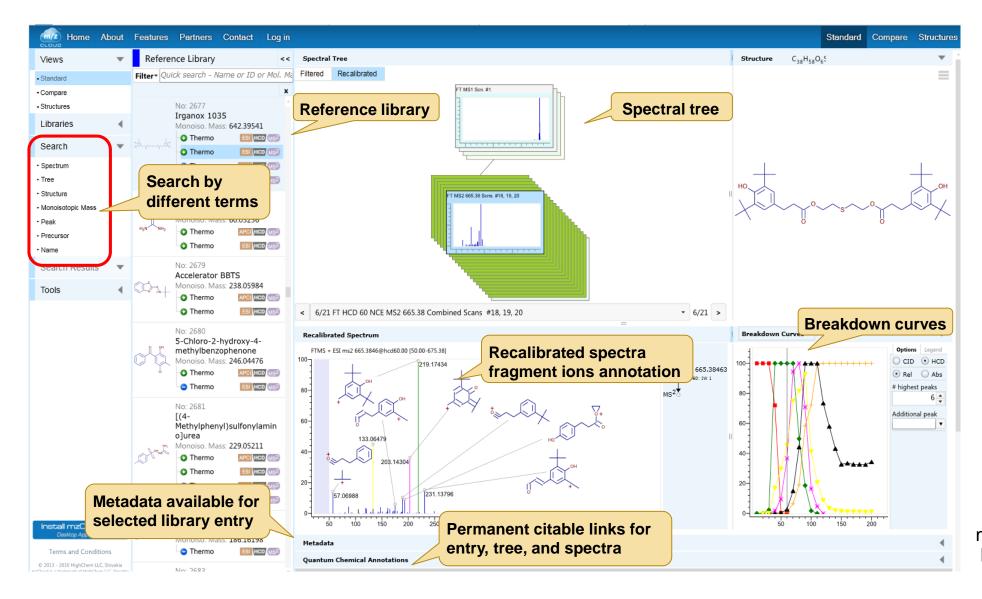
- -Advanced high resolution mass spectral database
- -Very high quality data: using standardized acquisition, highly curated data
- -Identifying compounds, even when they are not present in the library, through substructure search



mzCloud is a trademark of HighChem LLC, Slovakia



mzCloud™ Spectral Library



mzCloud is a trademark of HighChem LLC, Slovakia

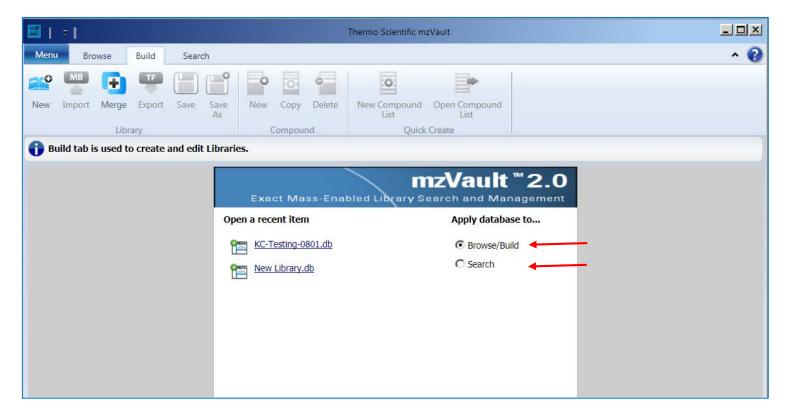


mzVault 2.0



Exact Mass-Enabled Library Search and Management

- Local mzCloud
- Creation of custom library
- Searching library





E&L Compound Database in Thermo Scientific Compound Discoverer 2.1

This Excel sheet database contains ~2000 common E&L related compounds. This is a "living document" and new E&L related compounds are added periodically.

Commercial Name	Chemical Name	Class	CAS No.	Chemical Formula	Formula weight	(M+H) ⁻	(M-H) ⁻	M•NH ₄ ·	M•Na [·]	M•K·	Structure	mzCLoud link
dioctyl phthalate	ethylhexylphthalate	PL	117-81-7	C24H38O4	390.27701	391,28429	389.26973	408.31083	413.26623	429.24017	>-\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\\	https://mzcloud.org/DataViewer.aspx#CReference2
Dibutyl phthalate	Dibutyl phthalate	PL	84-74-2	C16H22O4	278.1518	279.15908	277.14452	296.18562	301.14102	317.11496		https://mzcloud.org/DataViewer.aspx#CReference2
	Diogolohexyl phthalate		84-61-7	C20H26O4	330.18311	331.19039	329.17583	348.21693	353.17233	369.14627	d	https://mzcloud.org/DataViewer.aspx#CReference2
	□iethyl phthalate		84-66-2	C12H14O4	222.08921	223.09649	221.08193	240.12303	245,07843	261.05237		https://mzcloud.org/DataViewer.aspx#CReference2
	Dimethyl phthalate		131-11-3	C10H10O4	194.05791	195.06519	193.05063	212.09173	217.04713	233.02107		https://mzcloud.org/DataViewer.aspx#CReference2

This database has been added to Compound Discoverer as "E&L Mass List".



Thermo Scientific Compound Discoverer 2.1: Compound ID and Structure Elucidation

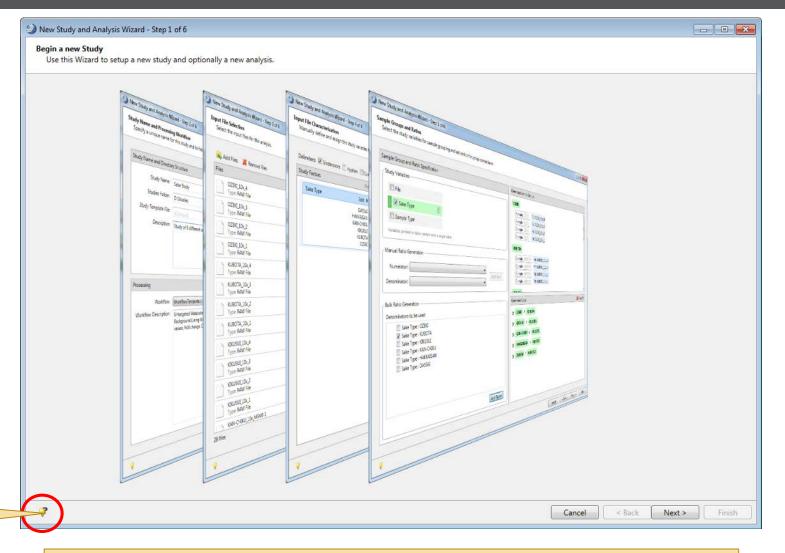
- Compound Discoverer is a small molecule structure analysis software package.
 - It processes high HRAM data generated by Thermo Scientific™ Orbitrap™ MS (LCMS & GCMS)



- It uses flexible node-based processing workflow and HRAM data, isotope pattern matching, conduct component extraction and elemental composition prediction. The component structures are identified using the following techniques:
 - Known compound ID through multiple database searching and fragment spectral matching.
 - Known unknown compound ID ChemSpider molecular formula and molecule weight search: using "Structure Proposals" feature to interrogate the fragmentation allows identification of the correct structure.
 - Total unknown structure elucidation For components without database search results, the predicted compositions, MS/MS fragment ions, and similarity search result (mzCloud fragment ions matches) are used to propose putative structure. De Novo structure ID
 - Validity check of the proposed structure Using "FISh Scoring" feature, searching the embedded "HighChem Fragmentation Library". FISh stands for "Fragment Ion Search"
- Differential analysis for different lots, batch, or type of sample analyses



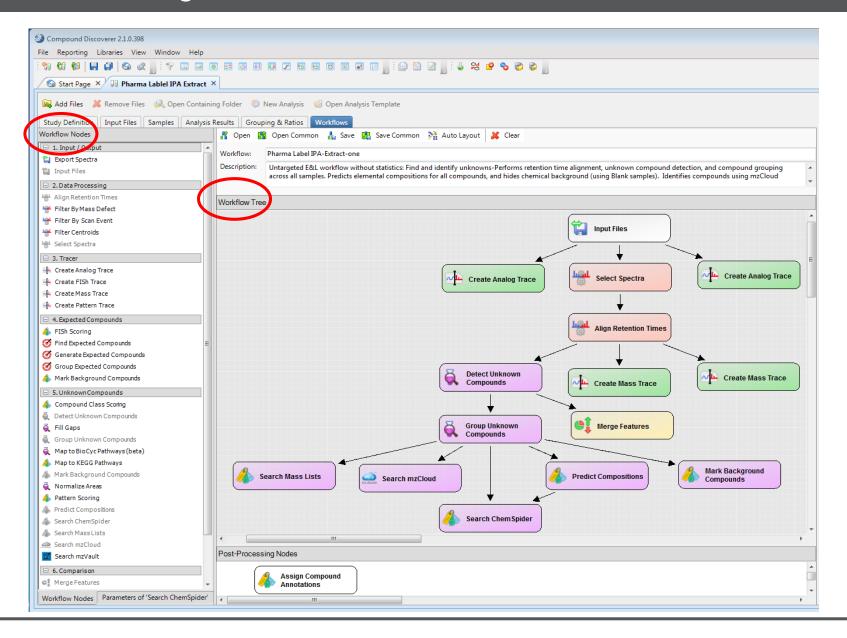
Thermo Scientific Compound Discoverer 2.1: Wizard for Processing Workflow Build



Description for each step

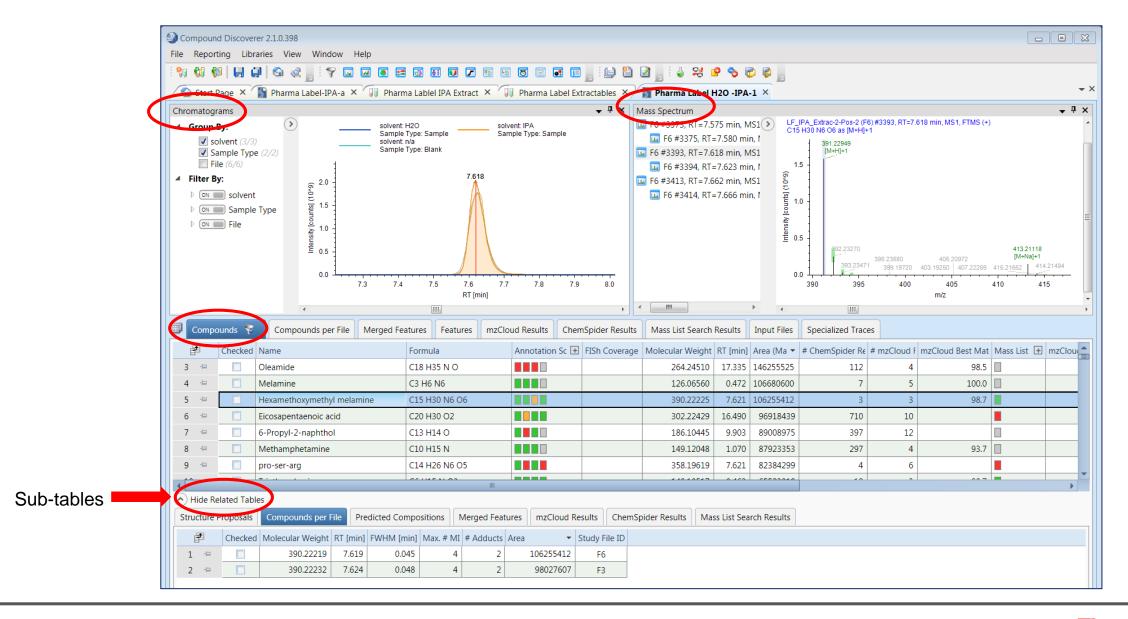
Creating a study and analysis using the guided "New Study and Analysis Wizard" and built-in workflow templates.

Node-Based Processing Workflow

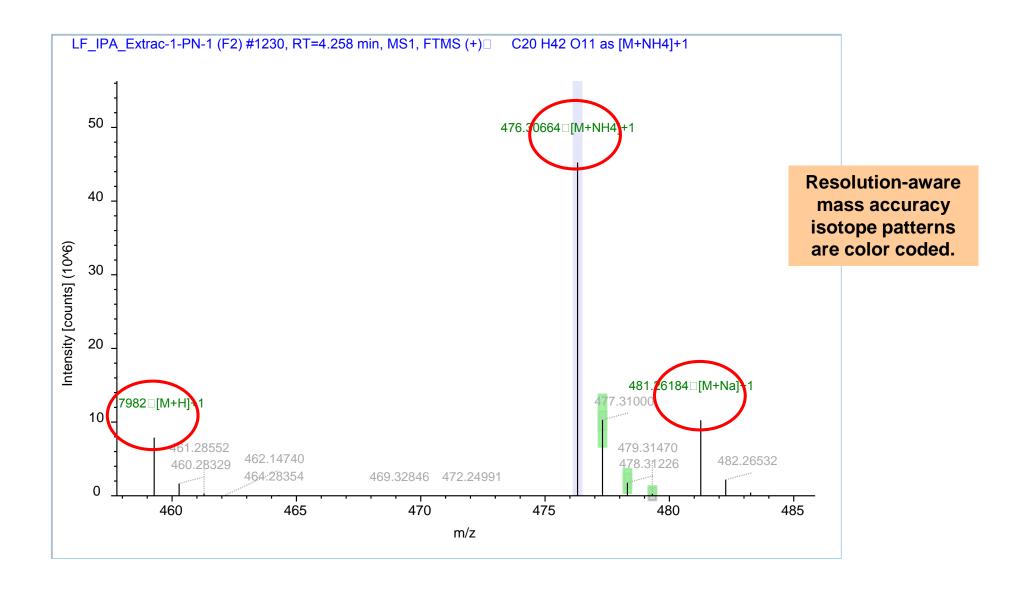




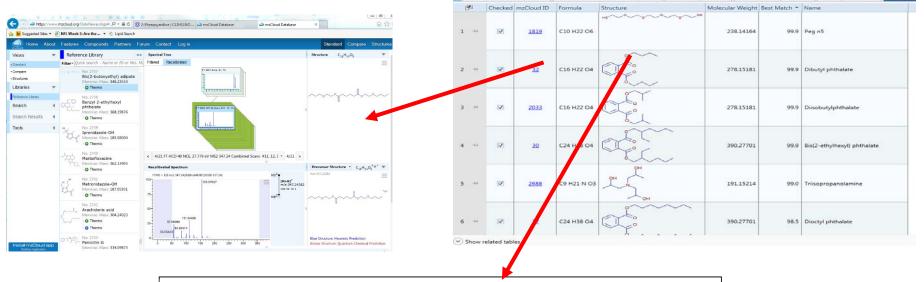
Result View – Data Interpretation

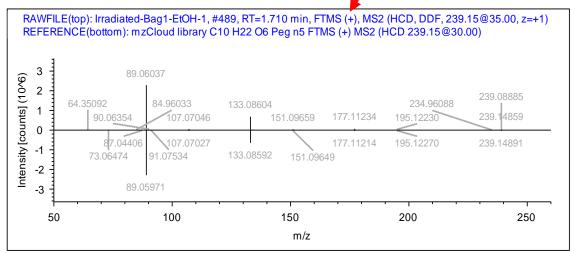


Component Detection with Adducts Grouping



Known Compound ID Through mzCloud™ Database Search



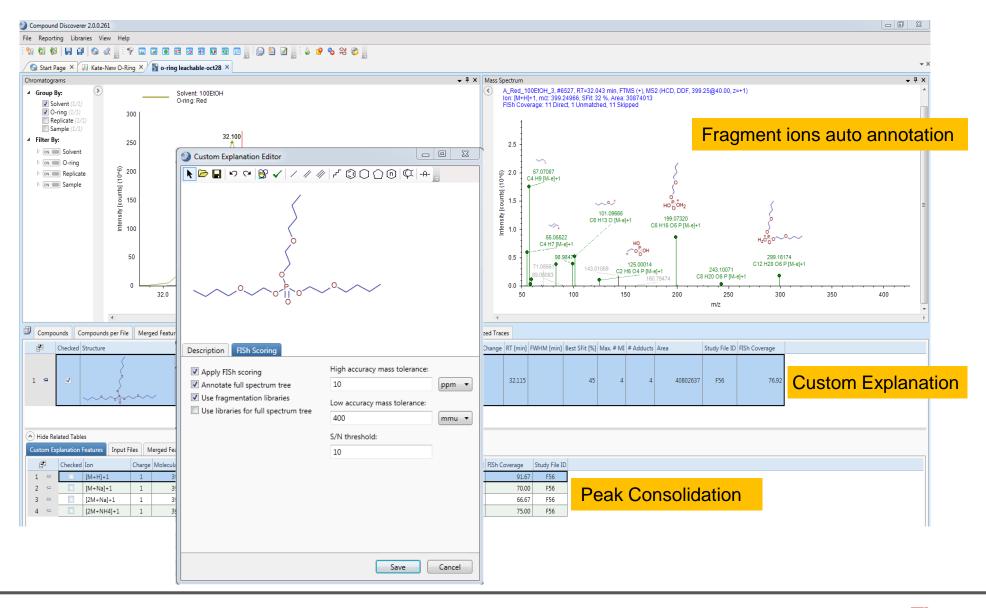


mzCloud is a trademark of HighChem LLC, Slovakia

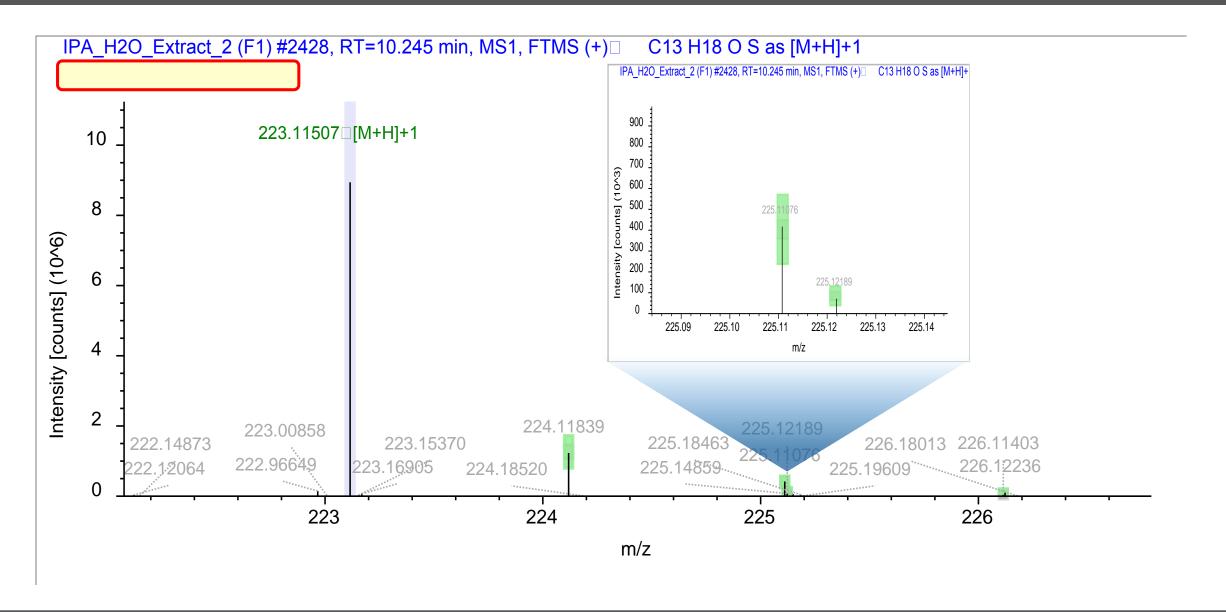
mzCloud™ Mirror plot



Unknown Structure Elucidation Using Compound Annotation Editor and FISh Scoring



Unknown Structure Elucidation Starting from CD 2.1 "Predicted Compositions"



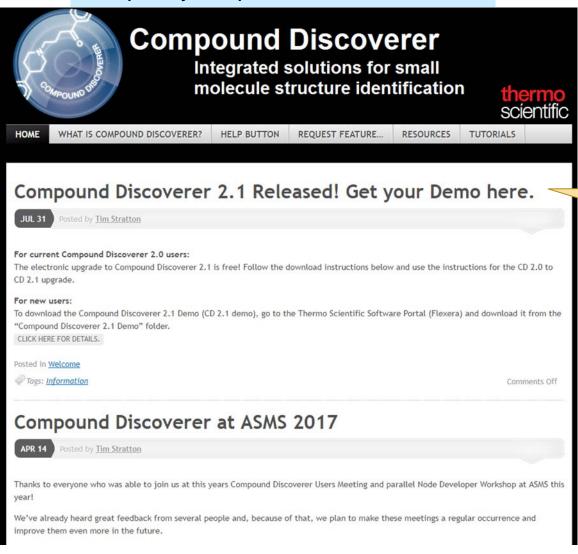
Thermo Fisher Compound Discoverer 2.1 Differential Analysis Feature: Volcano Plot

Using the interactive volcano plot to find compounds that are significantly different between two sample groups



Thermo Scientific Compound Discoverer (CD) Home Page

http://mycompounddiscoverer.com



Download a DEMO copy, and watch the tutorials videos here.

Join the Fun! Cache a Chromeleon Game

- Use your mobile device to complete challenges and earn a Charlie Chromeleon plush toy!
- If you are playing, you have earned points for attending this seminar. Be sure to scan the barcode on the desk outside the door.
- Ask booth staff for more details on how to play.



Please join me in the

Mass Spectrometry

section of our booth where I'll
address additional comments and questions.